

Characteristics of mineral elements in shoots of three annual halophytes in a saline desert, Northern Xinjiang

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Abstract: Halophytes are valuable salt-, alkali- and drought-resistant germplasm resources. However, the characteristics of mineral elements in halophytes have not been investigated as intensively as those in crops. This study attempted to investigate the characteristics of mineral elements for annual halophytes during their growth period to reveal their possible physiological mechanisms of salt resistance. By using three native annual halophytes (*Salsola subcrassa*, *Suaeda acuminata* and *Petrosimonia sibirica*) distributed in the desert in Northern Xinjiang of China, the dynamic changes in the mineral element contents of annual halophytes were analyzed through field sampling and laboratory analyses. The results demonstrated that the annual halophytes were able to absorb water and mineral nutrients selectively. In the interaction between the annual halophytes and saline soil, the adaptability of the annual halophytes was manifested as the accumulation of S, Na and Cl during the growth period and maintenance of water and salt balance in the plant, thus ensuring their selective absorption of N, P, K, Ca, Mg and other mineral nutrients according to their growth demand. By utilizing this property, halophyte planting and mowing (before the wilting and death periods) could bioremediate heavy saline-alkali soil.

Keywords: annual halophyte; mineral elements; desert; saline-alkali soil; Northern Xinjiang

Citation: Ke ZHANG, ChunJian LI, ZhongShao LI, FuHai ZHANG, ZhenYong ZHAO, ChangYan TIAN. 2013. Characteristics of mineral elements in shoots of three annual halophytes in a saline desert, Northern Xinjiang. Journal of Arid Land, 5(2): 244–254.

The saline land in Xinjiang accounts for about one third of all the saline land in China; approximately one third of the cultivated land in Xinjiang has undergone secondary salinization (Zhao and Li, 1999; Xi et al., 2006a). Vegetation coverage in saline deserts is very low, and increasing vegetation coverage can contribute greatly to reducing the greenhouse effect. However, the most promising aspects of increasing vegetation coverage are the change in soil fertility and the ecological improvement of the organisms' living environment.

In the soil-plant system interaction in saline deserts, the soil affects the characteristics of chemical elements in plants, which are closely related to the soil salt properties and react to soil fertility, for example,

both *Tamarix* spp. and *Haloxylon ammodendron* Bge., under their canopies, show fertile island (or salt island) features (Xi et al., 2006b; Li et al., 2007; Yin et al., 2008a). Halophytes, grown in saline soil, have a higher soluble salt content (Na^+ , Cl^-) (Flowers et al., 1977; Osmond et al., 1980; Ungar, 1991; Xia, 1994; Donovan et al., 1997; Yin et al., 2002; Zhang et al., 2007; Gao et al., 2010; Umethan et al., 2012). Previous studies were conducted on the characteristics of chemical elements in different halophytes in various habitats; they focused on the analysis of total salt, as well as Cl^- , Na^+ and SO_4^{2-} (Xia, 1994; Yin et al., 2002; Zhang et al., 2007; Gao et al., 2010), and the relation of salt and mineral nutrients (Marschner, 1995; Bai et al., 2008; Kudo and Fujiyama, 2010; Yuan et al., 2010).

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Received 2012-08-22; revised 2012-10-16; accepted 2012-11-02

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The contents of minerals in halophytes are not in proportion to those in saline soil, and the contents of some elements (Cl, Na) are far greater than those in glycophytes, which have different plant nutrition mechanisms with halophytes. However, there is little available literature on the macro and medium nutrient elements in halophytes, especially the dynamics of nutrient elements during the growth period, and their relationship with element concentrations in the soil. The growth properties of annual halophytes are similar to those of crops. Studies on plant nutrition regulation in saline soil using annual halophytes as the model plants (Flowers and Colmer, 2008) will provide an important basis for studying plant nutrition in saline agriculture.

1 Study area

The study area is located in the primary saline desert region of the National Field Observation and Experiment Station for Desert Ecology in Fukang, the Chinese Academy of Sciences. This area is to the northern side of Bogda Peak of Tianshan Mountains, and is an oasis/desert ecotone of the south Gurbantunggut Desert, belonging to the lower part of the Sangonghe alluvial plain. The area is located at 87°56'E and 44°17'N, with an elevation of 475 m, a typical temperate desert climate, an annual precipitation of 100–200 mm (70%–80% of which falling between May–September), and an annual evaporation of 1,000–2,000 mm. Groundwater level in the study area is 2.9–5.3 m (changing with seasons) and has a salinity of approximately 2 g/L. Drought, high temperature, and strong light bring about seasonal surface-soil salt accumulation, leading to the generation of sulfate-dominant saline soil in which native succulent shrubs and herbs are sparsely scattered. This kind of habitat is commonly called a succulent halophytic desert. The total annual sunshine hours are 2,532.5 h, and the frost-free period is 174 d (Xu and Li, 2006; Xu et al., 2007).

2 Materials and methods

Three native annual halophytes distributed in the saline desert of Northern Xinjiang, i.e. *Salsola sub-*

crassa, *Suaeda acuminata* and *Petrosimonia sibirica*, were chosen as the research subjects (Zhang et al., 2012). Samples were collected in the primary saline desert at Fukang station.

Sampling sites, which are 100 m² in area, were randomly chosen in the primary saline desert where the three native halophytes were distributed. The sampling sites were flat, being far away from shrubs and under little human disturbance, so there were enough materials for sampling at different time points during the whole growth period. The halophyte shoots were obtained at different growth periods in 2009 and 2010, respectively. Halophyte shoots and soil in 0–100 cm soil profiles were sampled on different dates in 2009 and 2010. Three 30 cm×30 cm quadrats were randomly selected as 3 replicates, with a total of 9 quadrats for the three halophyte species; ditches of 100-cm depth were dug under plants in each quadrat; soil samples for each 10-cm soil layer from bottom to top were collected for laboratory analysis. Plant counts were made for each quadrat: all plant shoots were cut and washed rapidly, air-dried, and freshly weighed, and then bake-dried at 105°C and weighed, crushed, and finally sieved for later use.

Determination of soil samples included: available N, P and K; water-soluble Ca, Mg, Na, Cl and S. Determination of plant samples included: total N, P, K, Ca and Mg in the shoots; water-soluble Na, Cl and S. Succulence was expressed as the ratio of water content in the shoots: (Fresh weight–Dry weight)/Fresh weight.

Available soil N, P and K were determined using the Distillation-Kelvin Digestion method, Olsen-P method, and Ammonium Acetate extraction method; water-soluble salts (Ca²⁺, Mg²⁺, Na⁺, Cl[–] and SO₄^{2–}) were determined using conventional methods (1:5 soil/water ratio). After sampling solutions were obtained using the Sulfuric acid–Perchloric acid digestion Method, each element (N, P, K, Ca and Mg) was measured as follows: N by Azotometer-Kelvin Digestion method, P by Mo-Sb-colorimetry, K, Na, Ca and Mg by Atomic Absorption Spectrometry (Bao, 2000). Plant water-soluble salts (Na⁺, Cl[–] and SO₄^{2–}) were measured by conventional titration after boiling water extraction (Zhou et al., 2000).

Data processing was performed using Excel 2003 and SAS 8.1 one-way ANOVA.

3 Results

3.1 Soil properties of saline desert

The analysis of the soil properties in 2009 and 2010 showed that soil water content, pH and bulk density increased from top to bottom. On the contrary, soil organic matter content was the highest in surface soil layer, and decreased from top to bottom. The electric conductivity of soil solutions of different layers in the two years differed only slightly, except for a relatively high value in surface soil layer in both years (Zhang et al., 2012).

3.2 Changes of shoot water content during the growth period

Shoot water contents changed only slightly in the three plant species during the growth period (Fig. 1). The variation ranges of average water content between different sampling periods in 2009 and 2010 were 830–870 and 680–800 g/kg, respectively. Inter-annual variation was greater than the variation between different periods of each year. The data from both years demonstrated that, during the early growth period, the succulence of *S. acuminata* was the highest. With the

extension of growth period, the succulence of *S. subcrassa* became the highest; the second and third highest were with *S. acuminata* and *P. sibirica*, respectively. At the last harvest in 2010, the three plant species withered, and the shoot water content was very low.

3.3 Changes of N, P, K, Ca and Mg concentrations in different soil layers during the growth period

3.3.1 Available soil N

Data from both years showed that the change in available soil N (expressed as the mean for the 1-m soil profile), according to the time scale, was as follows: the available N began to increase in spring, reached its peak in June, and then decreased. The spatial distribution of the change in available N concentration was as follows: during the early growth period, it was relatively high in the middle and lower soil layers; during the middle and late growth periods, it was highest in the surface soil layer and evenly distributed below the surface soil layer, but fluctuated in the middle and lower soil layers during the different periods (Fig. 2).

3.3.2 Available soil P, K, water-soluble Ca and Mg

The vertical distribution of available soil P, K, and water-soluble Ca gradually decreased from top to bottom, except that the available P decreased in the 0–40 cm layer and was stable in the lower layers; water-soluble Ca demonstrated a transitional increase in the middle soil layer (Figs. 2, 3). The water-soluble Mg concentration in the soil gradually increased from the surface layer to lower layers, but gradually decreased in the 60–80 cm soil layer (Fig. 2). The temporal distribution characteristics of the above elements (according to the mean for the 1-m soil profile) were as follows: during the growth period, the available P concentration reached its peak from late May to early July in both years, suggesting that temperature and precipitation imposed certain influences on this period when the maximum available P concentration occurred. The available K concentration changed slightly in the growth season; it maintained a value of about 0.335 g/kg in 2010, and fluctuated within the range of 0.259–0.318 g/kg in 2009. Water-soluble Ca concentration increased during the early growth period, and then decreased to a minimum value in September,

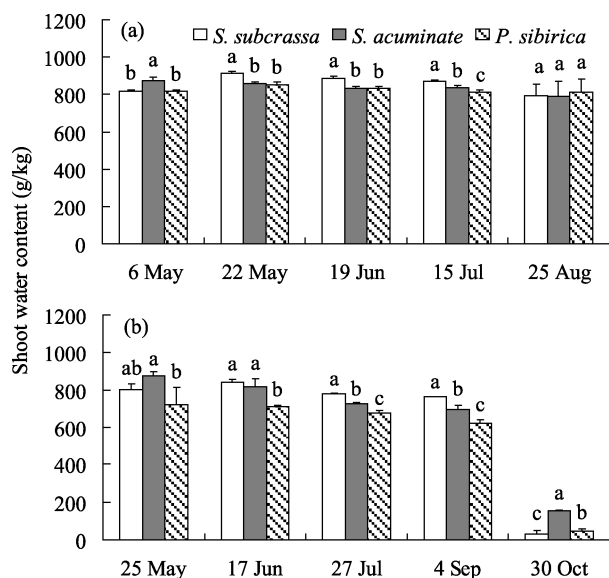


Fig. 1 Changes of shoot water content during the growth periods in 2009 (a) and 2010 (b). Different letters above columns for each sampling date represent significant difference among plant species ($P < 0.05$). Means \pm SE, $n=3$.

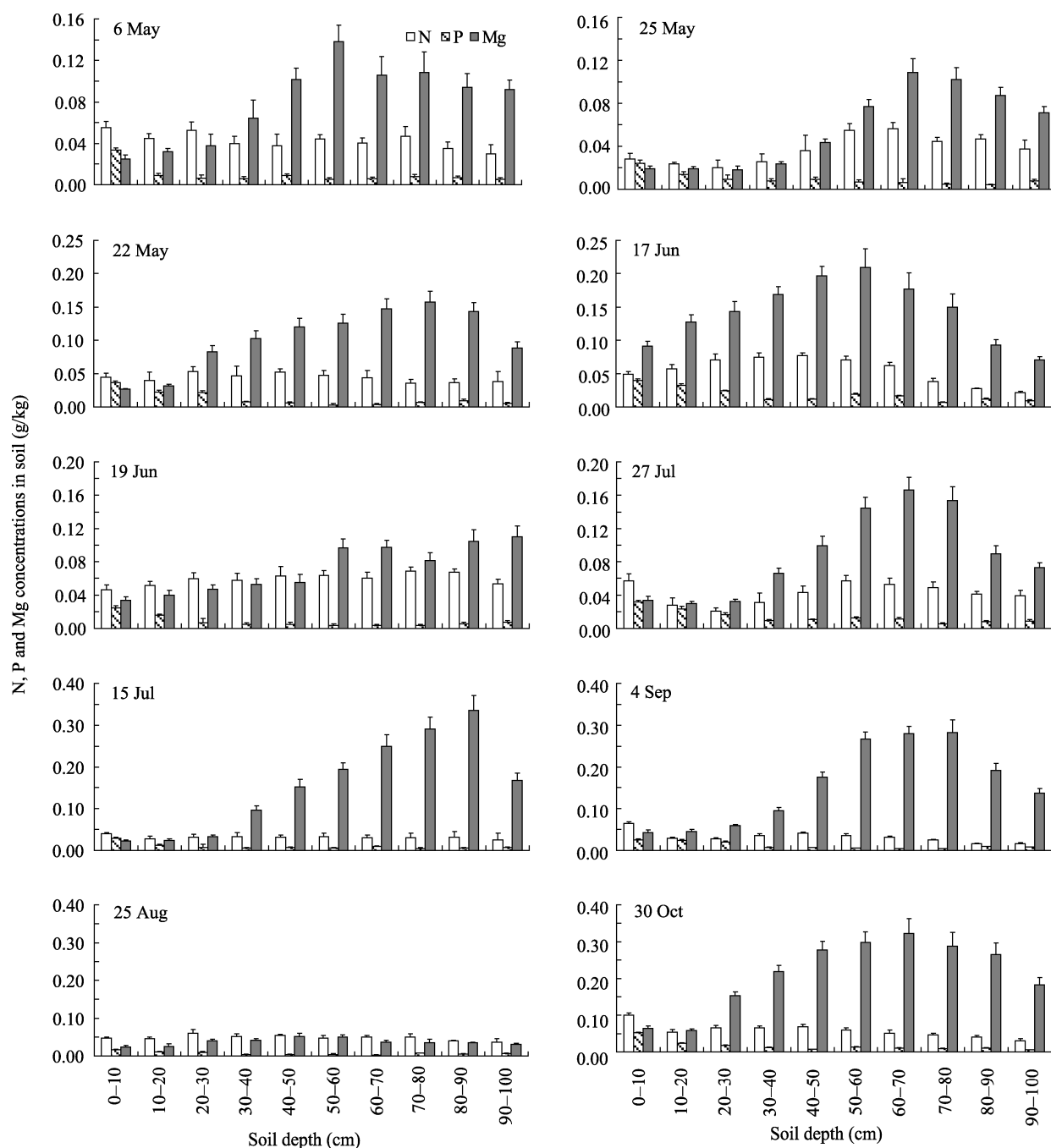


Fig. 2 Changes of available N, P and water-soluble Mg concentrations in the 0–100 cm soil profile during the growth periods in 2009 (left column) and 2010 (right column)

and increased again in November. The water-soluble Mg concentration reached its peak in July 2009 and September 2010; the change curve of 2010 lagged one month behind that of 2009.

3.4 Changes of Na, Cl and S concentrations in different soil layers during the growth period

The characteristics of the spatial distribution of avail-

able Na^+ , Cl^- and SO_4^{2-} in the soil (Figs. 3 and 4) were as follows: the distribution of Na^+ and SO_4^{2-} in different soil layers in the two years was similar, developing from surface accumulation during the early period to vertically even distribution; the contents in each soil layer below the surface soil layer were similar, distributed evenly. Cl^- gradually increased from the

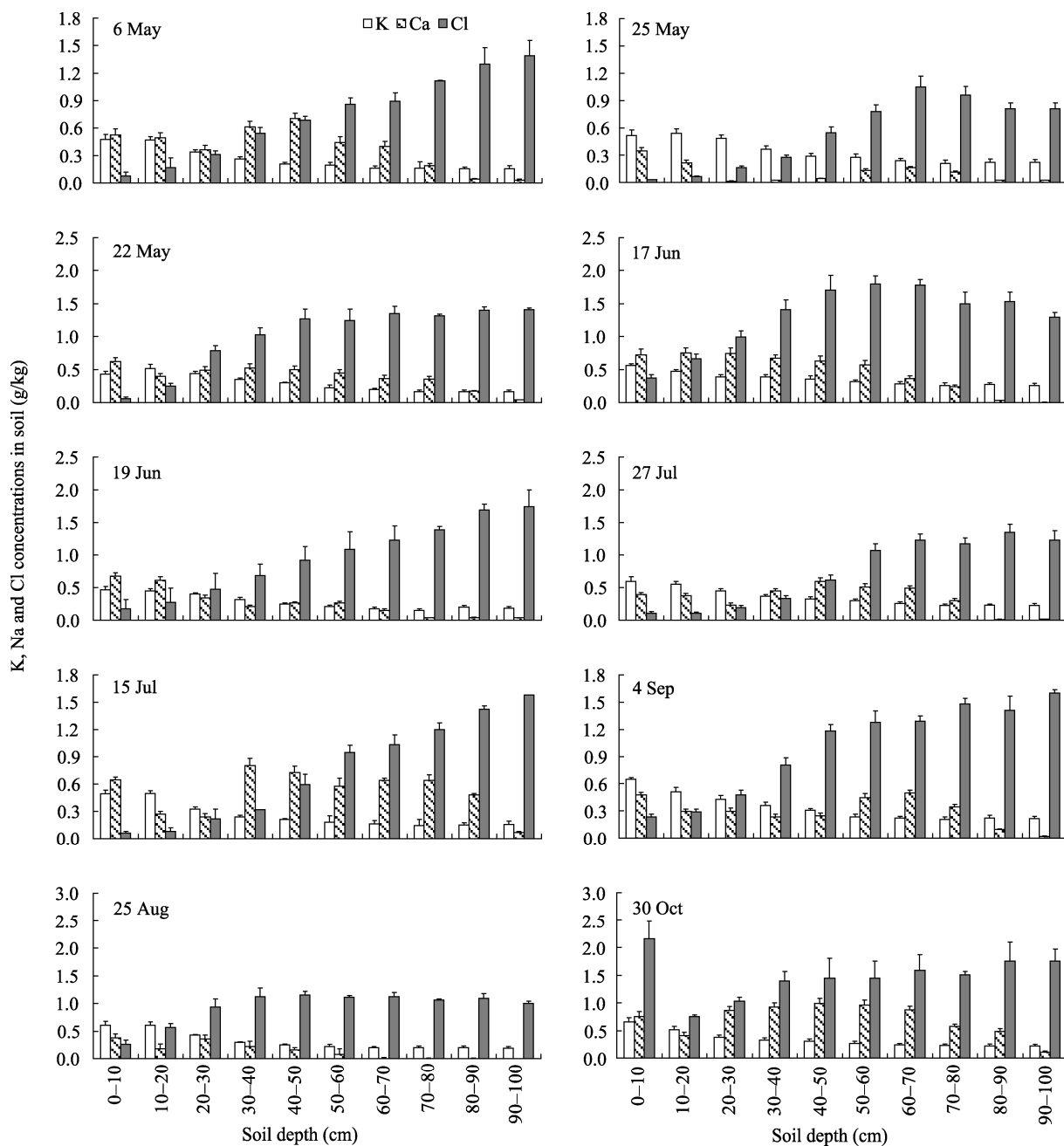


Fig. 3 Changes of available K, water-soluble Ca and Cl concentrations in the 0–100 cm soil profile during the growth periods in 2009 (left column) and 2010 (right column)

surface layer to deeper layers. The temporal distribution characteristics were as follows: Na^+ and SO_4^{2-} concentrations decreased slightly with the extension of growth period in 2009, and on the contrary, they increased with the extension of growth period in 2010. Cl^- exhibited an increasing tendency with the extension of growth period in both years. The concentration of soil SO_4^{2-} was the highest; the second highest concentrations were with Na^+ and Cl^- . N and P concentrations were the lowest, suggesting that the main cations

in saline soil were Na^+ , Ca^{2+} , Mg^{2+} and K^+ , and the main anions were SO_4^{2-} and Cl^- .

3.5 Changes of N, P, K, Ca and Mg concentrations in the shoots

N, P and K concentrations in the shoots of the three annual halophytes showed the same characteristics: N concentration was the highest; the second highest was with K; and P concentration was the lowest (Fig. 5). All of them gradually decreased with the advance of

growth period, but the variation ranges were small. The order of N, P and K concentrations in the shoots was *S. acuminata* > *S. subcrassa* > *P. sibirica*.

A comparison of the Ca and Mg concentrations of the three halophytes during each sampling period in the two years indicated that the water-soluble Ca

concentration was lower than the water-soluble Mg concentration in most cases. The order of the water-soluble Ca concentration was *P. sibirica* > *S. acuminata* > *S. subcrassa*, and the order of the water-soluble Mg concentration was *P. sibirica* > *S. subcrassa* > *S. acuminata*.

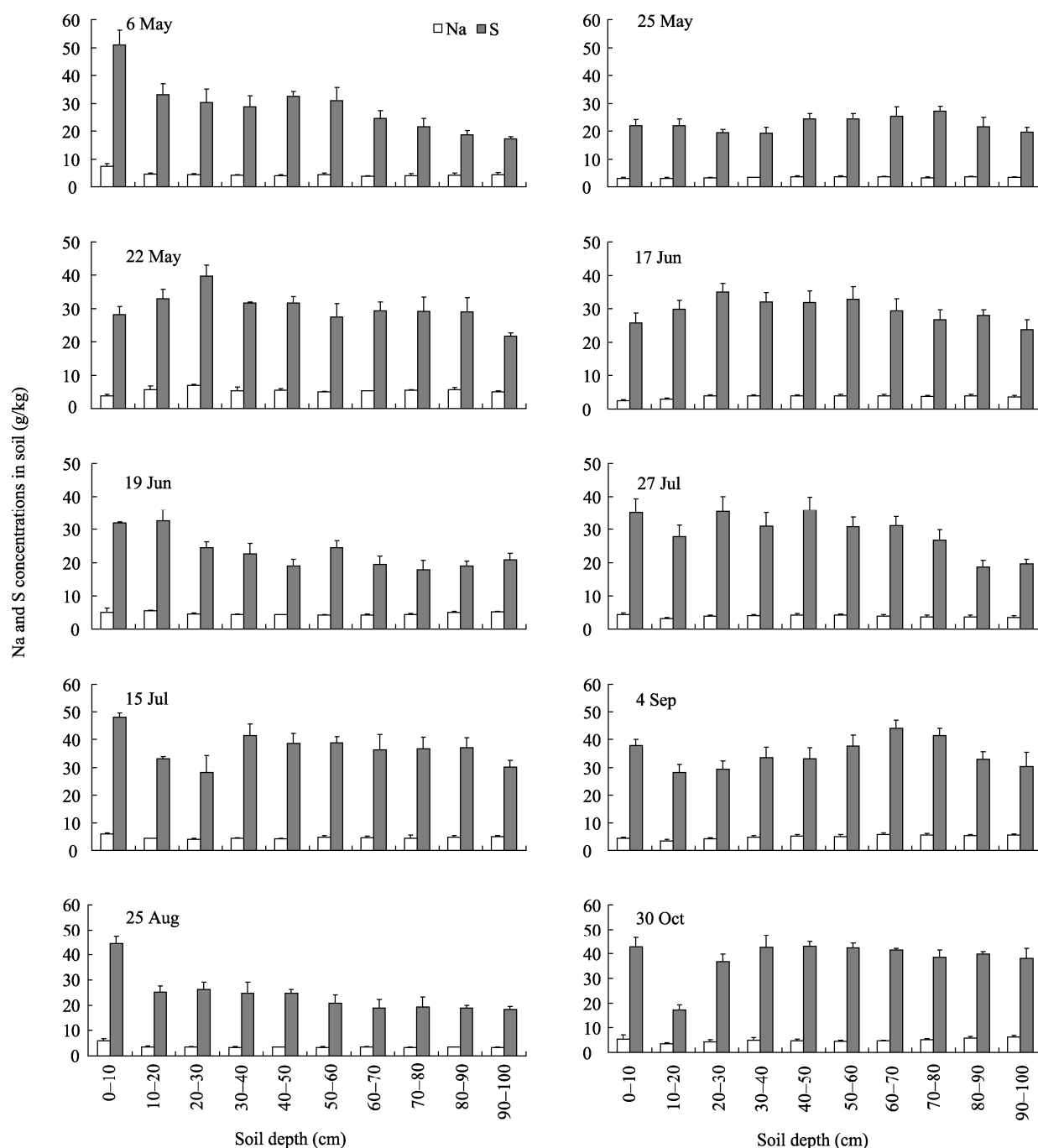


Fig. 4 Changes of water-soluble Na and S concentrations in the 0–100 cm soil profile during the growth period in 2009

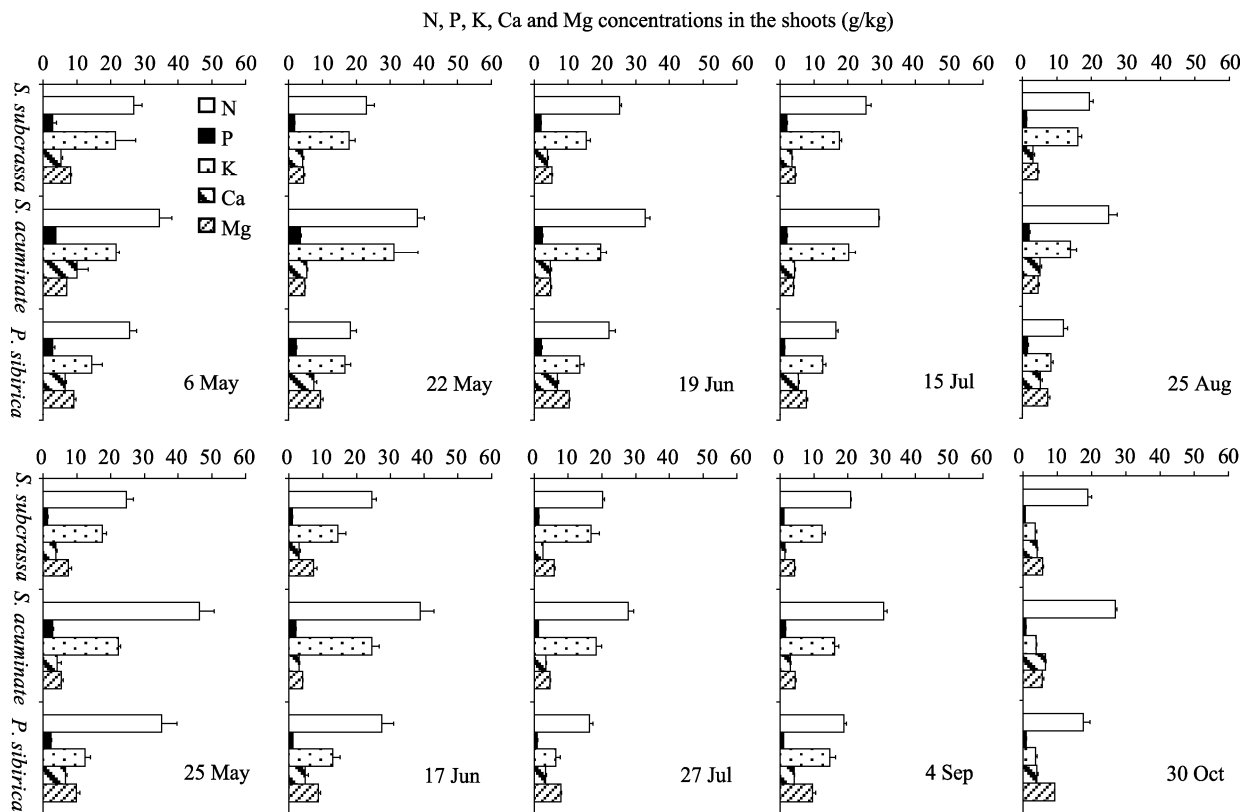


Fig. 5 Changes of N, P, K, Ca, Mg concentrations in the shoots of the three annual halophytes during the growth periods in 2009 (upper row) and 2010 (lower row)

3.6 Changes of Na, Cl and S concentrations in the shoots

Na, Cl and S concentrations in the shoots of the three annual halophytes showed the same characteristics: S concentration was the highest and Cl concentration was the lowest (Fig. 6). In 2009, Na concentration in the shoots showed a rapidly increasing tendency during the course of the growth period, and that of *S. subcrassa* increased most rapidly. In 2010, Na concentration increased during the early period, reaching its peak in June, and then decreased slightly, tending to be stable; Na concentration decreased significantly at the last harvest (death period). The order of Na concentration in the shoots of the three halophytes was *S. subcrassa* > *P. sibirica* > *S. acuminata*.

In 2009, the data for water-soluble Cl and S concentrations in the shoots were only obtained for *S. subcrassa* (Fig. 6). Data from both years showed that the water-soluble Cl concentration in the shoots of *S. subcrassa* displayed an increasing tendency with the extension of growth period, and maintained stable

after reaching a certain level. Cl concentration in the shoots of the three plant species decreased to the minimum at the last harvest (death period). The order of Cl concentration in the three halophytes was *S. acuminata* > *S. subcrassa* > *P. sibirica*.

In 2009, the water-soluble S in *S. subcrassa* increased continuously with the extension of growth period, but the case was not the same in 2010. Similar to those of Na and Cl, S concentration in the three halophytes decreased significantly at the last harvest in 2010. The order of S concentration was *S. subcrassa* > *P. sibirica* > *S. acuminata*. The ratios of S, Na and Cl concentrations in the three halophytes were 5:2:1, 4:2:1 and 1:1:1 for *S. subcrassa*, *P. sibirica* and *S. acuminata*, respectively.

4 Discussion

4.1 Characteristics of water content in the shoots of the three annual halophytes

Water content is of special significance to annual halophytes. In the arid regions of Xinjiang, where

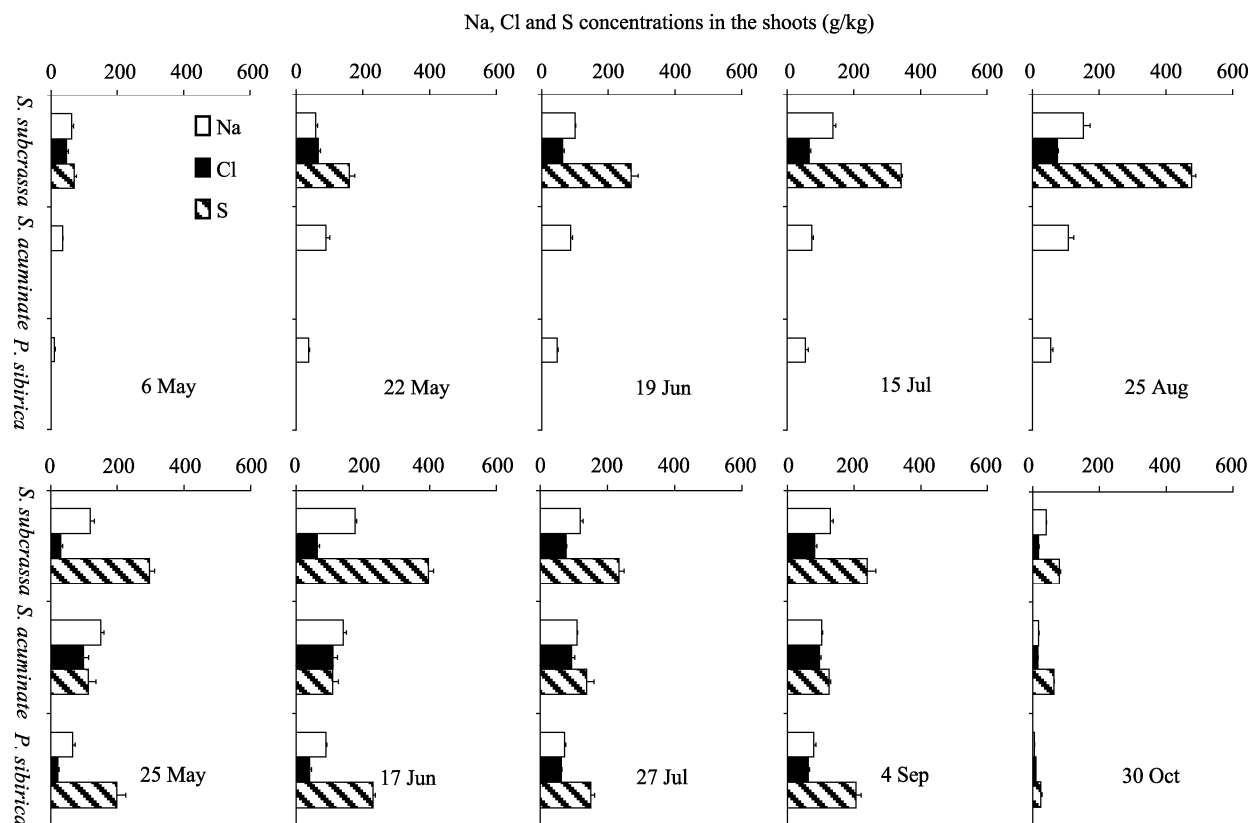


Fig. 6 Changes of water-soluble Na, Cl and S concentrations in the shoots of the three annual halophytes during the growth periods in 2009 (upper row) and 2010 (lower row)

precipitation is rare and soil salinity is high, halophytes have developed a leaf/stem succulence structure similar to desert plants. Succulence reduces leaf surface area, which is beneficial for reducing evapotranspiration and damage from wind and sand; therefore it is clear that the higher the succulence, the stronger the stress resistance (Liu et al., 2006). Shoot water content was adopted to represent succulence in this study; it reflects not only succulence, but also plant water content. Comparison of the three annual halophytes revealed that the order of succulence was *S. subcrassa* > *S. acuminata* > *P. sibirica*. The differences in succulence also reflect different salt-drought resistant capacities to some extent. The water content of pasture plants is generally 590–800 g/kg (Lai et al., 2008). It is only in the active parts of crops such as the stem shoots, tender stems and roots that the water content can reach above 800–900 g/kg. The water content in the shoots of the three annual halophytes was maintained at 800 g/kg throughout the entire growth period. Huang (1988) demonstrated that suc-

culent plants (tissue water content ranging from 750–800 g/kg) had a low evaporation ratio, while non-succulent plants (tissue water content ranging from 500 to 650 g/kg) had a high evaporation ratio. Therefore, maintaining relatively high water content during the entire growth period is one of the key strategies for the annual halophytes to adapt to extreme climatic and physiological drought.

4.2 Characteristics of macro and medium element concentrations in the soil and shoots

Soil in the desert of Northern Xinjiang was demonstrated to be chloride-sulfate saline soil, where the content of SO_4^{2-} was about 28 times that of Cl^- and the main composition of salt was Na_2SO_4 (Figs. 3 and 4). Cation concentrations in the soil were similar ($\text{Ca}^{2+} > \text{K}^+ > \text{Mg}^{2+}$), and they were 0.09, 0.07 and 0.03 times that of Na^+ , respectively. As macro mineral nutrients, although N and P were far lower than salt (Figs. 2 and 4), their contents reached mid-levels compared with the case in other soils, suggesting that the saline

soil in the desert of Northern Xinjiang is not infertile and its productivity is affected by salinity rather than by nutrition. Due to the movement of groundwater (especially capillary water), concentrations of available soil N and water-soluble Mg gradually increased from the surface layer to lower layers, which is one of the features of saline soils in arid regions.

The concentrations of elements S, Na and Cl were the highest in the three annual halophytes, and were similar to the order of the elements in the soil. Soil salinity affected the absorption of the three elements. Na, Cl and S concentrations in crops were generally 0.01–80, 0.01–80 and 1.5–15 g/kg, respectively (Epstein and Bloom, 2005), while those in the annual halophytes reached 10–150, 41–110 and 70–477 g/kg, respectively. Thus, it is clear that S, Na and Cl concentrations were the highest in the annual halophytes (much higher than the concentrations of other mineral elements), which was related to the highest concentrations of the three elements found in soil (Figs. 3–6). According to data for the period of strong growth (27 July), the order of salt content in the three annual halophytes was *S. subcrassa* > *S. acuminata* > *P. sibirica*. Generally, plants with higher salt content have greater salt tolerant capacities (Fig. 6). On the other hand, the high concentrations of S, Na and Cl in the halophytes were clearly the basis for the plants' ability to obtain and sustain water in arid saline deserts; only by maintaining a low water potential with high salinity, could a series of physiological activities be carried out (Flowers et al., 1977; Gorham et al., 1980; Yeo, 1983). The concentrations of water-soluble Na, Cl and S were very low in the plants on 30 October, 2010 (during the wilting and death period), suggesting that salt was not bound and could be discharged along with water loss. Since it was only during wilting and death period that salinity decreased significantly, it is possible that salt was discharged mainly through the leakage led by plasma membrane rupture. The salt was then distributed onto the surface of the plants with water and scattered in the surface soil with litter, leading to salt bioaccumulation in the surface soil layer. Our findings were supported by previous studies, which reported that, along with plant death, Na evaporated to the plant surface and scattered in the surface soil with litter, resulting in salt accumulation

in this layer of the soil and the saline islands of halophytic shrubs such as *Tamarix chinensis* (Li et al., 2007; Yin et al., 2008b).

The order of N, P, K, Ca and Mg concentrations in the three annual halophytes was $N > K > Mg > Ca > P$, which was different from that in soil where the order was $Ca > K > Mg > N > P$. Saline soil modified the nutrition characteristics of the three annual halophytes, and as a result, they not only accumulated a lot of salt, but also absorbed a large amount of N for plant growth. This situation not only ensured the demand for nutrition absorption, but also helped to maintain low water potential (Flowers et al., 1977; Gorham et al., 1980; Yeo, 1983; Kudo and Fujiyama, 2010). Mineral element concentrations in the annual halophytes growing in saline soil were passively affected by soil element contents and also actively influenced by the selective absorption of different elements. The ratio of selective absorption was expressed as the element concentration ratio of plant to soil.

Taking Na for example, element absorption ratio (AR) could be expressed by the following equation:

$$AR_{Na} = Plant_{Na} / Soil_{Na}$$

$Plant_{Na}$ is the concentration of Na in the plant, and $Soil_{Na}$ is the concentration of Na in the soil. Absorption ratio represents the elementary uptake of plant to soil. The absorption ratios of N, P, Cl, Mg, K, Na, Ca and S were calculated to be approximately 558, 152, 62, 57, 49, 20, 12 and 7, respectively. It can be seen that the contents of N, P, Mg and K were significantly increased in the annual halophytes, but were much lower in the soil; however, the selective absorption ratios of macro elements (Na and S) in the soil were relatively small, suggesting that the annual halophytes were able to control salt absorption. This selection mechanism requires further study.

4.3 Ratios of different elements in the three annual halophytes

Available N, P and K, water-soluble Ca, Mg, Na, Cl and S in the soil can be regarded as the mineral components which are easily absorbed and utilized by plants. As can be seen from the ratios of different mineral elements in Table 1, the ratio of N in the annual halophytes was far higher than the ratio in soil and in crops as well. This suggests that halophytes have a high N absorption capacity. The ratios of S, Na

and Cl were the highest in saline soil and in the three annual halophytes as well, but their ratios were far lower in plants than in soil, demonstrating the plants' selectivity in salt absorption. The ratio of Ca in saline soil was high, while Ca accumulation was very low in the annual halophytes. This is related to the succulent structure of the three halophytes, which can decrease evaporation and reduce water loss.

In comparison of the three annual halophytes, there is little difference in the ratios of most nutritional elements, except for a relatively large difference in the ratios of Na, Cl and S. According to the results in Table 1, *S. subcrassa* and *P. sibirica* can be referred to as Na₂SO₄-type plant and *S. acuminata* can be referred to as NaCl-type plant. Different absorption of mineral elements may reflect interspecific differences.

Table 1 Comparison of the concentration ratios of different elements in halophyte shoots, saline soil and crops

	Concentration ratio (N:P:K:Ca:Mg:Na:Cl:S)
<i>S. subcrassa</i>	16:1:11:2:4:82:41:63
<i>S. acuminata</i>	15:1:9:2:2: 41:37:17
<i>P. sibirica</i>	13:1:7:3:5:31:24:33
Mean for the three annual halophytes	15:1:9:3:4:49:34:34
Saline soil	4:1:27:34:10:381:84:783
Crops (Epstein and Bloom, 2005)	(3–12):1:(5–16):(0.7–12):(0.3–2):(0.007–16):(0.007–16):(1–3)

Note: in the soil, N, P and K are available nutrients, and Ca, Mg, Na, Cl and S are water-soluble elements; in the shoots, N, P, K, Ca and Mg are total elements, and Na, Cl and S are water-soluble elements. Data for the plants are the mean values of different sampling periods in 2009 and 2010; data for the soil are the mean values in the 1-m soil profile of different sampling periods in both years. Concentration of S was converted from that of SO₄²⁻ (S concentration=SO₄²⁻ concentration×(S/SO₄)=SO₄²⁻ concentration×(32/(32+16×4)).

5 Conclusions

Maintaining high water content during the entire growth period is one of the important strategies for the annual halophytes to adapt to extreme climatic and physiological drought.

N, P, K and Ca concentrations in the annual halophytes gradually declined with the advance of the growth period, then maintained within a certain range, and eventually decreased slightly during the wilting and death period (except for the Ca concentration, which significantly increased during the death period). The water-soluble Mg concentration maintained with-

in a certain range throughout the entire growth period, with only slight changes. By using Mg nutrient absorbed in the vegetative period, the annual halophytes can maintain the dynamic fluidity of Mg during the whole growth period.

Na, S and Cl concentrations were dominant elements in the annual halophytes, and were maintained at a high level during the growth period. Although the three annual halophytes accumulated a lot of salt during the growth period, salinity decreased significantly during the wilting and death period. By utilizing this property, planting halophytes and mowing them before the wilting and death period could effectively bioremediate heavy halomorph soil.

It can be concluded from the ratios of N, K, Na, Cl and S that, although the mineral element contents in the soil influence the distribution of mineral elements in the plants, the element composition in halophytes does not entirely depend on salt composition and quantity in the soil. This is due to their selective absorption properties and the differed adsorption of mineral elements among species.

Acknowledgments

This study was financially supported by the Scientific and Technological Project of Xinjiang Uygur Autonomous Region (201130106-2) and the Innovation and Sustainable Development Research on Forest Carbon Sink in Karamay. We would like to thank PiZhen HUANG, ChuanHua YIN, Lei WANG, Ping WANG, Li LI, LinXia WANG and MingFang HU for their advice and assistance. Thanks also go to Hui ZHANG and other laboratory staff.

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